



CANDIDATE NAME

CENTRE NUMBER

# UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Level

CANDIDATE NUMBER	



BIOLOGY 9700/53

Paper 5 Planning, Analysis and Evaluation

October/November 2012

1 hour 15 minutes

Candidates answer on the Question Paper.

No Additional Materials are required.

#### **READ THESE INSTRUCTIONS FIRST**

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black ink.

You may use a pencil for any diagrams, graphs or rough working.

Do not use staples, paper clips, highlighters, glue or correction fluid.

Answer both questions.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [ ] at the end of each question or part question.

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1		
2		
Total		

This document consists of 8 printed pages.



1 The single- celled alga *Chlorella* is common in polluted waters and is used in sewage lagoons to reduce the nitrate content of the water.

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Fig. 1.1 shows cells of *Chlorella* viewed with a light microscope using high power.

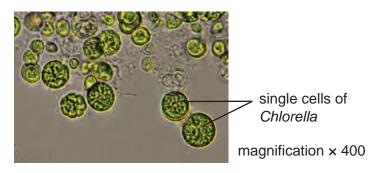


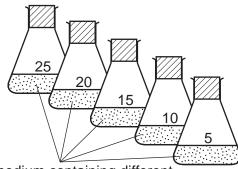
Fig. 1.1

A student investigated the effect of different concentrations of nitrate on the population growth of a species of *Chlorella*. The student used sodium nitrate in the investigation.

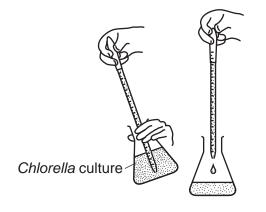
Fig. 1.2 shows the main steps in the procedure.

# Step 1 Step 2

- 250 cm<sup>3</sup> of a growth medium was added to each of five flasks
- a different concentration of sodium nitrate
   solution from 5 mmol dm<sup>-3</sup> 25 mmol dm<sup>-3</sup>
   was added to the flasks
- 1 cm<sup>3</sup> sample removed from a *Chlorella* culture using a graduated pipette
  - the 1 cm<sup>3</sup> sample transferred to a flask



growth medium containing different concentrations of sodium nitrate



### Step 4

- immediately the number of cells of Chlorella in 5 samples taken from each flask was counted.
- 5 samples were removed from each flask at intervals of 2 days for a period of 10 days and the number of cells counted.
- Step 2 repeated for the rest of the flasks

Step 3

 the flasks were left in sunlight at room temperature.

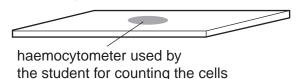


Fig. 1.2

a) (i)	Identify the independent and dependent variables in this investigation.	For
	independent	Examiner's Use
	dependent[1]	
(ii)	Identify <b>two</b> variables that the student has controlled in this investigation as shown in Fig. 1.2.	
	[1]	
(iii)	There are other variables that the student could have controlled in this investigation.	
	Describe how two other variables could have been controlled.	
	[4]	
b) (i)	The student used solid sodium nitrate to prepare the highest concentration of sodium nitrate solution shown in Fig. 1.2 (25 mmol dm <sup>-3</sup> ). This concentration was then used to prepare all the other concentrations. Describe the procedure that the student used to prepare the concentrations shown in Fig. 1.2. Your description should be sufficiently detailed so that another person can easily follow your procedure. The molar mass of sodium nitrate is 85 g mol <sup>-1</sup> .	
	[4]	

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(ii) The	student also prepa	ared anothe	r flask to use	e as a contro	l.	
Sug	gest a suitable sol	ution to use	as a control	for this inves	stigation.	
						[1]
(c) The resu	Its of the student's	investigatio	n are shown	n in Table 1.1		
		Tabl	e 1.1			
concentration of sodium	mean number of cells in	mean ¡	percentage	increase in <i>Chlorella</i>	number of o	cells of
nitrate / mmol dm <sup>-3</sup>	1 cm <sup>3</sup> at the start	day 2	day 4	day 6	day 8	day 10
control	10035	1	2	2	2	2
5	10630	40	75	140	175	220
10	10128	50	105	160	230	285
15	10743	75	135	210	292	360
20	10108	65	125	195	280	330
25	10900	55	110	170	250	300
(i) State	e how the mean p	ercentage ir	ocrease in nu	umber of cell	s was calcul	ated.
	gest why the stud			ı percentage		
						[1]

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	[3]
Suggest how the student's investigation should be modified to find the optimum nit concentration for the growth of this <i>Chlorella</i> .	trate
<u> </u>	
	[1]
Sewage lagoons also contain high concentrations of phosphate. In another investiga	
the student used sodium phosphate instead of sodium nitrate to find out the effective of the concentrations on the growth of <i>Chlorella</i> .	01 01
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Dialysis is used to treat kidney failure. One type of dialysis, called CAPD, allows patients to move around during treatment. During this type of dialysis a fluid containing the sugar dextrose, a form of glucose, is passed into the abdominal cavity and left for a few hours. Wastes and excess water pass into the dialysis fluid, which is then removed and thrown away.

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Fig. 2.1 shows dialysis fluid being added to the abdominal cavity.

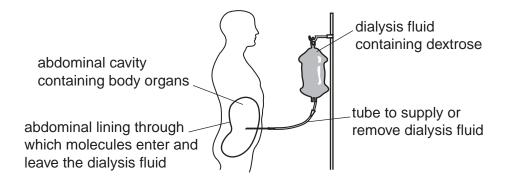


Fig. 2.1

(a) Tests are carried out on each patient receiving CAPD to find out if the treatment is effective. One of these tests uses strips that change colour to indicate the urea concentration.

Outline urea.	how te	st strips	could b	e used	to find	the e	ffectivene	ss of C	APD in	removi	ng
											[2]

One effect of kidney failure is anaemia which can be treated by injecting a hormone that stimulates red blood cell production. This treatment can cause a lack of iron used for the synthesis of haemoglobin. Iron supplements are usually taken by mouth.

Ten male and four female patients with a mean age of 63, all using dialysis and being treated for anaemia, took part in a study into a different way of giving iron supplements.

Test **A** was carried out on samples of blood from the patients at the start. Measurements were made of :

- haemoglobin content
- iron content
- ferritin, a plasma protein for transporting iron.

The patients were supplied with an iron supplement which they took by mouth for three months. At the end of this time test **B** was carried out on samples of blood from the patients by measuring the same three substances as before.

In the next part of the study the patients' iron supplements were replaced by an iron-dextran complex that was injected into their muscles once a week for three months. Test **C** was carried out on samples of blood from the patients in the same way as in tests **A** and **B**.

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# Table 2.1

	me	range of			
substance tested	test A (before treatment with iron supplements)  test B (after injectio with hormon and iron taken mouth)		test <b>C</b> (after injection with hormone and iron given by injection)	concentrations in people without anaemia	
haemoglobin / g dm <sup>-3</sup>	90 ± 1	100 ± 1	130 ± 3	120 – 130	
iron / μg dm <sup>-3</sup>	740 ± 15	540 ± 11	940 ± 21	500 – 1760	
ferritin / μgdm <sup>-3</sup>	2.81 ± 0.69	2.67 ± 1.04	6.60 ± 1.04	11 – 300	

(b) (i)	State what $S_M$ shows about the data.
	[2]
(ii)	A number of $t$ -tests were carried out to determine if the treatments with iron supplements had a significant effect on the concentrations of the substances tested. The results of test $\bf A$ were used as a comparison for both test $\bf B$ and test $\bf C$ . State why the $t$ -test is a suitable statistical test to use for the data.
	[1]
(iii)	State evidence in Table 2.1 that suggests the results of these <i>t</i> -tests might be significant.
	[1]

Question 2 continues on page 8

(c)	The researchers who carried out this study concluded that providing iron supplement by injection was more effective than providing iron by mouth.					
	(i)	State the evidence from Table 2.1 for this conclusion.				
	(ii)	Suggest limitations of this study.				
		[3]				
		[Total: 9]				

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