



**General Certificate of Education (A-level) Applied
January 2013**

Applied Science

SC05

**(Specification
8771/8773/8776/8777/8779)**

Unit 5: Choosing and Using Materials

Final

Mark Scheme

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Question	Part	Marking guidance		Mark	Comment
1	(a)(i)	strength	(AO1)	1	
1	(a)(ii)	stiffness	(AO1)	1	
1	(a)(iii)	brittleness	(AO1)	1	
1	(b)	undergoes plastic deformation / deforms permanently / will not return to original length (or shape) when force removed	(AO1)	1	
1	(c)(i)	mass per unit volume / mass ÷ volume	(AO1)	1	
1	(c)(ii)	2 from: balance displacement can measuring cylinder	(AO1) (AO1)	2	NOT: 'scales' alone - must be qualified accept eureka can
1	(c)(iii)	mass = density × volume mass of aluminium = $2.75 \times 80 = 220\text{g}$ mass of lead is also 220g volume of lead = $220 \div 11 = 20(\text{cm}^3)$	(AO2) (AO2) (AO2)	3	3 marks for correct answer (allow answer obtained by ratio) 2 compensation marks for: <ul style="list-style-type: none"> • $M = D \times V / V = M \div D /$ correct substitution of either • mass = 220g
2	(a)	<ul style="list-style-type: none"> • nylon • glass reinforced plastic(GRP) • tile 	(AO1) (AO1)	2	2 marks if all three answers are correct 1 mark if 1 or 2 answers are correct
2	(b)(i)	a long chain molecule / a long chain of monomers / a long chain of repeating units	(AO1)	1	do not accept 'a long chain <u>of</u> molecules'
2	(b)(ii)	made of more than one material	(AO1)	1	

2	(c)	alloy	(AO1)	1	
2	(d)(i)	can be hammered (or pressed) into shape	(AO1)	1	do not accept 'can be shaped'
2	(d)(ii)	<ul style="list-style-type: none"> in iron the layers / atoms can slide past each other the different sized atoms / irregular structure (of steel) prevents the layers / atoms sliding past each other (as easily) 	(AO1) (AO1) (AO1)	3	The second marking point can be obtained from a diagram
2	(e)	stainless steel does not corrode (or rust) / ordinary steel corrodes (rusts) / stainless steel is resistant to chemicals / stainless steel does not react with water	(AO1)	1	ignore aesthetic qualities
3	(a)	<ul style="list-style-type: none"> elastic deformation (to push refill back into pen) the spring must return to its original length 	(AO2) (AO2)	2	
3	(b)(i)	<ul style="list-style-type: none"> correct label and units added to each axis (1) all points plotted correctly (to within $\pm \frac{1}{2}$ a small square) (1) straight line of best fit through points and origin (1) 	(AO2) (AO2) (AO2)	3	
3	(b)(ii)	3.2(N)	(AO1)	1	allow 3.1 to 3.3(N)
3	(b)(iii)	new force = $\frac{1}{2}$ of original force / 1.6N (allow ecf)	(AO2)	1	
4	(a)(i)	man made / not natural	(AO1)	1	
4	(a)(ii)	any 2 from: <ul style="list-style-type: none"> lightweight waterproof / doesn't absorb so much water more flexible / soft / stretchy leather changes shape 	(AO2) (AO2)	2	ignore ethical comments

4	(b)	metal or polymer or composite (no mark) <u>metal</u> : 2 from: <ul style="list-style-type: none"> • strong / doesn't break • durable / tough / long lasting / hard wearing • hard <u>polymer</u> : 2 from: <ul style="list-style-type: none"> • lightweight • durable / long lasting / hard wearing • strong <u>composite</u> : <ul style="list-style-type: none"> • 2 from a combination of above 	(AO2) (AO2)	2	no marks for reasons if haven't stated a material or have stated ceramic / wood 2 correct materials each with 1 reason = 2 marks
4	(c)	any 3 from: <ul style="list-style-type: none"> • low density / lightweight • strong / doesn't break • durable / long lasting / hard wearing • hard • does not corrode • rigid / stiff / not flexible 	(AO2) (AO2) (AO2)	3	
5	(a)	can be drawn out into wires (or pipes) / shows (both elastic and) plastic behaviour (or deformation)	(AO1)	1	
5	(b)	<ul style="list-style-type: none"> • electrons are delocalised / free in structure / sea of electrons • these electrons can move in an organised way (or WTTE) / electrons carry the charge through the material 	(AO1) (AO1)	2	
5	(c)(i)	the higher the copper content, the higher the thermal conductivity / as one increases so does the other	(AO1)	1	accept converse accept positive correlation NOT: proportional

5	(c)(ii)	equation or description i.e. heat conducted per second ÷ (cross-sectional area × temperature gradient) or rate of flow of heat per unit area per unit temperature gradient	(AO1)	1	
5	(c)(iii)	<ul style="list-style-type: none"> zinc is the harder metal the higher the zinc content the harder the brass 	(AO2) (AO2)	2	
5	(d)	In any order: <ul style="list-style-type: none"> annealing (1). Heat to high temperature and cool slowly / cool in air / allow to cool / controlled cooling (1) quenching (1). Heat to high temperature and cool quickly / cool in water / cool in oil (1) 	(AO1) (AO1) (AO1) (AO1)	4	
5	(e)(i)	increase in temperature of 80°C causes 25cm to expand 0.025mm increase in temperature of 80°C causes 1cm to expand 0.001mm increase in temperature of 1°C causes 1cm to expand 0.001 ÷ 80mm =0.0000125(mm) or 1.25×10^{-5}	(AO2) (AO2)	2	2 marks for correct answer 1 compensation mark for correct answer to intermediary stage
5	(e)(ii)	coefficient of linear expansion / linear expansivity	(AO1)	1	
5	(e)(iii)	<ul style="list-style-type: none"> any correct example e.g. metal bridges (1) description e.g. bridge on rollers + expansion gap (2) 	(AO1) (AO1) (AO1)	3	for description allow either 2 precautions or 1 precaution plus explanation
6	(a)	a plastic made from plant products	(AO1)	1	
6	(b)	<ul style="list-style-type: none"> decomposes / decays / rots monomer turned into polymer 	(AO1) (AO1)	2	

6	(c)	less of the traditional plastics are needed / less crude oil is used to make traditional plastics	(AO1)	1	
6	(d)	lactic acid	(AO1)	1	
6	(e)(i)	covalent (bonding)	(AO1)	1	NOT covalent
6	(e)(ii)	electrons are shared	(AO1)	1	if 'ionic' is given in (d)(i) allow 'electrons are transferred' / 'force of attraction between ions'
6	(f)	any one of: <ul style="list-style-type: none"> • both are long chain molecules • both contain carbon • both contain hydrogen 	(AO1)	1	
6	(g)	any one of: <ul style="list-style-type: none"> • PLA contains oxygen, (poly(ethene) doesn't) • PLA contains double (covalent) bonds, (poly(ethene) contains only single (covalent) bonds) • PLA contains a CH₃ group 	(AO1)	1	accept 'it' for PLA
7	(a)(i)	strain = extension ÷ original length	(AO1)	1	
7	(a)(ii)	stress = force ÷ cross-sectional area	(AO1)	1	
7	(b)(i)	any 2 of: <ul style="list-style-type: none"> • micrometer screw gauge • vernier callipers • digital vernier • travelling microscope • metre rule • balance 	(AO1) (AO1)	2	

7	(b)(ii)	<ul style="list-style-type: none"> diameter of wire mass / load / force extension of wire <u>original</u> length of wire 	all 4 scores 2 marks 2 or 3 scores 1 mark	(AO3) (AO3)	2	
7	(b)(iii)	any 4 from: <ul style="list-style-type: none"> diameter using a micrometer in several places and calculate average value weight on a balance several weights used extension using vernier scale or travelling microscope original length using metre rule repeat readings on <u>unloading</u> 		(AO3) (AO3) (AO3) (AO3)	4	
7	(b)(iv)	any 4 from: <ul style="list-style-type: none"> force / load is given by mg diameter gives area using $\pi d^2 \div 4$ (or diameter gives radius which gives area using πr^2) for each value of force / load, calculate stress and strain plot a graph of stress against strain (or force against extension) gradient of graph = Young modulus (or $YM = \{\text{gradient} \times \text{length}\} \div \text{area}$) 		(AO3) (AO3) (AO3) (AO3)	4	
7	(b)(v)	hazard / risk identified and appropriate precaution given e.g. snapping wires and safety glasses e.g. falling weights and toe protection / sand bucket / stand back		(AO1)	1	

7	(c)	$\text{stress} = 35 \div 1.64 \times 10^{-7} = 2.13 \times 10^8$ $\text{strain} = 1.25 \times 10^{-3} \div 1.84 = 6.79 \times 10^{-4}$ $\text{YM} = \text{stress} \div \text{strain} = 2.13 \times 10^8 \div 6.79 \times 10^{-4}$ $= 3.14 \times 10^{11} \text{ Nm}^{-2} \text{ (Pa)}$ <p>3 marks for correct answer 1 mark for units</p>	<p>(AO2) (AO2) (AO2) (AO1)</p>	4	<p>2 compensation marks as follows:</p> <ul style="list-style-type: none"> • 1 mark for correct formula for Young modulus • plus 1 mark for correct value for either stress or strain • or 1 mark for correct substitution for either stress or strain <p>accept N/m^2 for units</p>
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