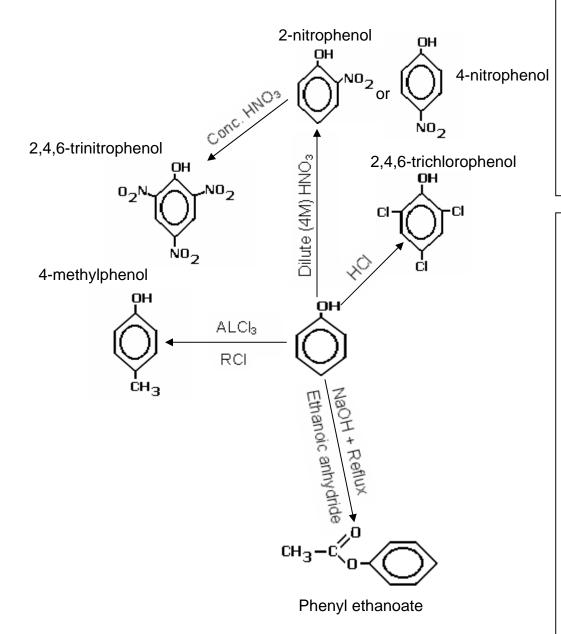
PHENOL



Uses of Benzene derivatives

Nitrobenzene and nitrophenol: Explosives

Dyestuffs

Sulphonic Acids: Drugs

Detergents Dyestuffs

Halogenophenol and benzene: Antiseptics

See Benzene Flowchart for relevant reactions.

Exceptions:

No Sulphuric Acid required for Nitration (3)

No Iron catalyst required for Halogenation (4)

Electrophilic Addition is much harder with Phenol due to increased stability provided by electron ring.

Esterfication:

Type: Substitution Mechanism: Nucleophilic

Nucleophile: Phenoxide ion $C_6H_5O^-$

Equations: $C_6H_5OH+Na^+ + OH^- \rightarrow Na^+ + C_6H_5O^- + H_2O^-$

$$cH^{3} - c > 0$$
 $cH^{3} - c > 0$
 $cH^{3} - c > 0$

Ethanoic Phenoxide Phenyl Ethanoate Anhydride Ion Ethanoate Ion