

**CAMBRIDGE INTERNATIONAL EXAMINATIONS**  
GCE Advanced Subsidiary Level and GCE Advanced Level

## **MARK SCHEME for the October/November 2012 series**

### **9701 CHEMISTRY**

**9701/34**

Paper 3 (Advanced Practical Skills), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components

|        |  |          |       |
|--------|--|----------|-------|
| Page 2 | Mark Scheme                            | Syllabus | Paper |
|        | GCE AS/A LEVEL – October/November 2012 | 9701     | 34    |

| Question | Sections  | Indicative material  | Mark |     |
|----------|---|--|------|-----|
| 1 (a)    | PDO<br>Recording  | <b>I</b> Correct units given for time and rates columns:<br>/ s or (s) and / s <sup>-1</sup> or (s <sup>-1</sup> )                                   | 1    | [9] |
|          |   | <b>II</b> Records all 5 times to the nearest second. Do not allow if t <sub>1</sub> > t <sub>3</sub> .   | 1    |     |
|          | PDO<br>Display  | <b>III All</b> (1000/time) values are correctly evaluated to 3 sig fig using the candidate's recorded times. (Minimum of 3 experiments carried out.) | 1    |     |
|          | MMO<br>Quality  | <b>IV to IX</b><br>Use the method given in the notes below when awarding the Quality marks.  | 6    |     |
|          | <p>Round all reaction times to the nearest second.</p> <p><b>IV and V</b><br/>Experiments 2 and 4: calculate <math>100(2t_2 - t_4)/t_4 \leq 20\%</math> for 1 mark; <math>\leq 10\%</math> for 2 marks.</p> <p><b>VI and VII</b><br/>Experiments 2 and 5: calculate <math>100(4t_2 - t_5)/t_5 \leq 20\%</math> for 1 mark; <math>\leq 10\%</math> for 2 marks.</p> <p><b>VIII and IX</b><br/>Experiments 4 and 5: calculate <math>100(2t_4 - t_5)/t_5 \leq 30\%</math> for 1 mark; <math>\leq 10\%</math> for 2 marks. If the candidate has not completed the 5<sup>th</sup> experiment, marks <b>IV</b> and <b>V</b> are available. Also check Experiments 1 and 2: t<sub>2</sub> should equal t<sub>1</sub> x 5/4. Use the 10% and 20% boundaries.</p> <p>If only the first three experiments are completed, award Q marks based on Experiments 1 and 2 (as above).</p> |  |      |     |
| (b)      | PDO<br>Layout   | <b>I</b> Plots (1000/time) on y-axis and volume of <b>FB 1</b> on x-axis. Axes correctly labelled and correct unit included with volume heading.     | 1    | [5] |
|          |   | <b>II</b> Uniform scales selected and more than half of the available grid used. Scales must start at (0,0).   | 1    |     |
|          |   | <b>III All</b> results are plotted within ½ square and in correct square. Allow for minimum 4 experiments carried out.                               | 1    |     |
|          |   | <b>IV</b> Draws a line through the origin (as shown) which lies within the arc of the points.  | 1    |     |
|          |   | <b>V</b> Draws a <b>straight</b> line of best fit (origin not essential).  | 1    |     |

| Page 3 | Mark Scheme                            | Syllabus | Paper |
|--------|--|----------|-------|
|        | GCE AS/A LEVEL – October/November 2012 | 9701     | 34    |

|     |                                       |   |                     |                    |
|-----|---------------------------------------|---|---------------------|--------------------|
| (c) | ACE Interpretation<br><br>PDO Display | (i) Experiment 1 and 5: correct concentration (to 2 – 4 sf) of hydrogen peroxide in one of the solutions (0.088/0.0885/0.08846 and 0.018/0.0177/0.01769 respectively).<br>Correct concentrations in both and working shown in one.<br><br>(ii) Working to show that concentration of H <sub>2</sub> O <sub>2</sub> is proportional to volume of <b>FB 1</b> .<br>Use of ratios or multiplying factor <b>or</b> statement that total volume is constant / the same in each.  | 1<br><br>1<br><br>1 | [3]                |
| (d) | ACE Conclusions                       | <b>Two</b> pieces of evidence needed.<br><b>If website statement correct</b><br>(i) a straight line / (line has) constant gradient<br>(ii) passes through origin <b>if</b> graph line is straight<br>(iii) straight line passes through origin (if appropriate from results) gains both marks.<br><b>or</b><br><b>If website statement not correct</b><br>(i) a curve has been drawn / no straight line / not constant gradient<br>(ii) straight line does not pass through the origin<br>(iii) points too scattered / not on best fit line.<br><br><b>If no comment on correct / incorrect</b><br>Allow 1 mark: for two pieces of evidence<br><br>A straight line, not passing through the origin could score both marks depending on explanation given (proportional but not directly proportional).<br>If two points are compared they must be on or very close to the graph line. | 1<br>1              | [2]                |
| (e) | ACE Conclusions                       | Predicts time will be reduced / halved (reference to rate is incorrect; allow time is faster).<br>Explains that smaller amount / moles / volume of thiosulfate are present to delay blue-black colour / less iodine needs to be produced.   | 1<br>1              | [2]                |
| (f) | ACE Interpretation                    | Temperature change / concentration of KI / initial concentration of H <sub>2</sub> O <sub>2</sub> . (NOT catalyst)  | 1                   | [1]                |
| (g) | ACE Interpretation                    | (i) Correctly calculates mean = <b>54.8 only</b> .<br>(ii) Correctly calculates error = <b>3.6</b> or <b>3.65%</b> .<br>Allow ecf correctly calculated from candidate's answer in (i) (3.56 or 3.6% if mean = 56.2).  | 1<br>1              | [2]                |
| (h) | ACE Improvements                      | 1 <sup>st</sup> experiment: only <b>FB 2</b> changes and distilled water adjusted to give 60 cm <sup>3</sup> total <b>and</b><br>2 <sup>nd</sup> experiment: only <b>FB 4</b> changes and distilled water adjusted to give 55 cm <sup>3</sup> total.  | 1                   | [1]                |
|     |                                       |   |                     | <b>[Total: 25]</b> |

| Page 4 | Mark Scheme                            | Syllabus | Paper |
|--------|--|----------|-------|
|        | GCE AS/A LEVEL – October/November 2012 | 9701     | 34    |

| FB 5 is FeSO <sub>4</sub> (aq); FB 6 is NH <sub>4</sub> Cl(aq) + Na <sub>2</sub> SO <sub>3</sub> (aq); FB 7 is MgSO <sub>4</sub> (aq); FB 8 is CH <sub>3</sub> CO <sub>2</sub> Na(s) |     |                    |   |            |     |
|--|-----|--------------------|---|------------|-----|
| 2  | (a) | PDO<br>Recording   | I Records all results (in correct space) for unknowns in a single table.  | 1          | [4] |
|  |     | MMO<br>Collection  | II Records green ppt, insoluble in excess NaOH for <b>FB 5 and</b> white ppt insoluble in excess NaOH with <b>FB 7</b> .  | 1          |     |
|  |     | MMO<br>Decisions   | III Only heats the solution in which no ppt formed with NaOH.   | 1          |     |
|  |     |                    | IV Tests <u>gas</u> /NH <sub>3</sub> evolved on heating <b>FB 6</b> with NaOH with (red) litmus paper turning blue.   | 1          |     |
|  | (b) | MMO<br>Collection  | With <b>FB 5</b> records a green ppt, insoluble in excess ammonia<br><b>and</b><br>with <b>FB 7</b> records a white ppt insoluble in excess ammonia.<br>Any evidence of the green ppt with <b>FB 5</b> turning brown in tests in (a) or (b).  | 1<br><br>1 | [2] |
|  | (c) | ACE<br>Conclusions | <b>No ecf in this section.</b><br><b>FB 5</b> contains Fe <sup>2+</sup> , iron(II)<br><b>FB 6</b> contains NH <sub>4</sub> <sup>+</sup> , ammonium<br><b>FB 7</b> contains Mg <sup>2+</sup> , magnesium   | 1          | [1] |
|  | (d) | MMO<br>Decisions   | (i) Chooses as reagents:<br>barium chloride / nitrate as <b>first</b> reagent,<br><b>and</b><br>hydrochloric / nitric acid as <b>second</b> reagent.  | 1          | [4] |
|  |     | MMO<br>Collection  | (ii) White ppt for all three with first reagent.<br>(Allow off-white ppt with <b>FB 5</b> )<br><b>FB 5</b> and <b>FB 7</b> ppt insoluble <b>and</b> <b>FB 6</b> ppt dissolves in second reagent.<br>(If acid added before Ba <sup>2+</sup> then award 3 <sup>rd</sup> mark for white ppt, no reaction, white ppt.)  | 1<br><br>1 |     |
|  |     | ACE<br>Conclusions | (iii) Correctly identifies the ions present<br><b>and</b><br>explanation from observations:<br>SO <sub>4</sub> <sup>2-</sup> in <b>FB 5</b> and <b>FB 7</b> as ppt insoluble in (appropriate) acid<br><b>or</b> SO <sub>3</sub> <sup>2-</sup> in <b>FB 6</b> as ppt soluble in acid.<br>(Only allow ecf if same transposition of solutions as in (a); SO <sub>3</sub> <sup>2-</sup> must be with NH <sub>4</sub> <sup>+</sup> ) | 1          |     |
|  |     |                    |   |            |     |

|               |   |                 |              |
|---------------|---|-----------------|--------------|
| <b>Page 5</b> | <b>Mark Scheme</b>                            | <b>Syllabus</b> | <b>Paper</b> |
|               | <b>GCE AS/A LEVEL – October/November 2012</b> | <b>9701</b>     | <b>34</b>    |

|            |                   |  |                   |     |
|------------|-------------------|--|-------------------|-----|
| <b>(e)</b> | MMO<br>Collection | <b>Either</b> solution turns yellow / orange / orange-brown / brown (box 1)<br><b>or</b> brown / rust / red-brown ppt formed (box 2)<br>(ppt soluble in excess is incorrect).<br>Other of the above <b>and</b> observes effervescence / fizzing / bubbles (in either box).<br>(Allow gas relights glowing splint (in either box) for 3 <sup>rd</sup> observation.) | 1                 | [2] |
|            |                   |  | 1                 |     |
| <b>(f)</b> | MMO<br>Collection | Test 1: (blue) litmus paper turns red<br><b>and</b><br>Test 2: sweet / fruity / glue / adhesive / nail varnish smell.<br>Accept smell of ester.  | 1                 | [2] |
|            | ACE<br>Conclusion | Salt of an organic / carboxylic acid or organic salt / named salt of organic acid<br><b>or</b><br>(A solid/crystalline) organic/carboxylic acid/named organic acid.  | 1                 |     |
|            |                   |  | <b>[Total:15]</b> |     |