



Mass of calcium ions =  $0.000805 \times 40 = 0.0322 \text{ g} = 32.2\text{mg}$   
 in  $25\text{cm}^3$   
 Mass of calcium ions in  $100 \text{ cm}^3$   
 Mass of calcium ions in  $100 \text{ cm}^3 = 32.2 \times 4 = 128.8\text{mg}$

Table comparing experimental and published data

Sample	From experiment	Published
Whole milk	128.8mg	119mg
Skimmed milk	133.6mg	124mg
Semi-skimmed	132mg	120mg
Long life semi skimmed	134.4 mg	120mg
Goats milk	132mg	120mg
Soya milk	141.6 mg	148mg
Tea milk	132mg	Not given
Flora	152mg	Not given
Yoghurt milk	114.1 mg	Not given

More variety of recording / processing and selection of data is needed to reach mark band 3

Further results of the investigation - could include

- a survey on daily intake of milk etc
- types of milk people drink etc
- need for calcium
- statistical data on milk intake (type/amount/cohort)

Possibly further experimental work on the contents of milk to include a different type of calculation /method of recording could also be added

Complex calculations etc. could involve error calculations for AO2c.

Data has been related to the objectives of the investigation

AO2 b Mark Band 1  
2 marks have been awarded.

Comments on differences in results needed to obtain Mark band 2 AO2b.

Also Mark band 2 could be awarded if the results of the weighings and the calculation for the reference solution containing  $\text{Ca}^{2+}$ .

**Total Marks AO2 = 6**